

An Efficient Synthesis of *anti*-(1R)-(+)-Camphorquinone 3-oxime.

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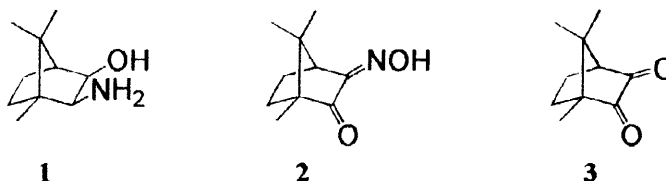
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Received 7 April 1998; accepted 23 June 1998

Abstract : Nitrosation of camphor with *iso*-amylnitrite in presence of base leads to the title compound in high yields. Excellent stereoselectivity is observed leading to preferential formation of the *anti*-isomer. © 1998 Elsevier Science Ltd. All rights reserved.

The camphor derived amino alcohol (**1**) is used as chiral auxiliary/reagent in various asymmetric transformations.¹ Reduction^{1b,g,2} of camphorquinone 3-oxime (**2**) appears to be the most common synthetic strategy^{1b,g,2} for **1**. However the poor yield for the preparation of **2** *via* the nitrosation³ of camphor led to the alternative protocols^{1b,d,2c} employing costly camphorquinone (**3**) as the starting material. Realising the poor yield during the reported nitrosation procedure³ to be a result of concomitant reduction of camphor by Na in EtOH or Et₂O, we planned to use nonreducing bases.⁴ Acid catalysed nitrosation procedures⁵ were avoided due to the possibility of Beckmann rearrangement of the resultant α -oximino ketone.^{3c,6} We report in this communication that **2** could be prepared in high yields (70-100 %) following the nitrosation route under various conditions. However, the *anti/syn* ratio was found to be dependent upon the base/solvent combinations⁷ and in most of the cases the *anti*-isomer was formed almost exclusively.⁸



In a typical experiment (1R)-(+)-camphor (7.61 g, 50 mmol) in dry THF (10 ml) was added dropwise to a magnetically stirred solution of ^tBuOK (1.73 g, 60 mmol) in dry THF (15 ml) at -30°C under nitrogen. The mixture was stirred at -30°C for 10 min., treated with ⁱAmONO (7.03 g, 60 mmol), and left at room temperature under stirring for 12hr. The solvent was distilled off, the residue diluted with water and extracted with Et₂O to separate any neutral component. The aqueous part was acidified with cold HOAc and extracted with Et₂O to afford the crude oxime as light yellow solid which on crystallisation (hexane - EtOH) afforded the desired oxime⁹ (7.7 g, 85 %).

References and Notes:

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- Yield (%) and *anti/syn* ratio: NaNH₂/THF/-30°C (90; 50:50); NaNH₂/Et₂O/-30°C (85; 66:34); NaH/Et₂O/-30°C (85; 98:2); NaH/DMF/-30°C (95; 100:0); KH/DMF/-30°C (90; 100:0); KH/THF/-30°C (70; 95:5); ^tBuOK/THF/-30°C (85; 100:0); ^tBuOK/PhH/5°C (100; 95:5); LDA/THF/-78°C (90; 95:5).
- Anti/syn* ratios of 3:2 (ref. 3) and 5:1 (ref. 1b) were reported earlier. Pure *anti* oxime was obtained by boiling the mixture several hours in water(ref. 6a) or through repeated crystallisation (ref. 6b).
- M.p. 150°C; [α]_D²⁵ +185° (c 4.5, CHCl₃); IR (KBr): 3400, 1740, 1640 cm⁻¹; ¹H NMR(300 MHz, CDCl₃) δ 3.25(d, 1H, *J* = 4.4), 2.07-2.01(m, 1H), 1.82-1.77(m, 1H), 1.60-1.52(m, 2H), 1.03(s, 3H), 1.00(s, 3H), 0.88(s, 3H); CIMS(m/z): 182(100, MH⁺).